

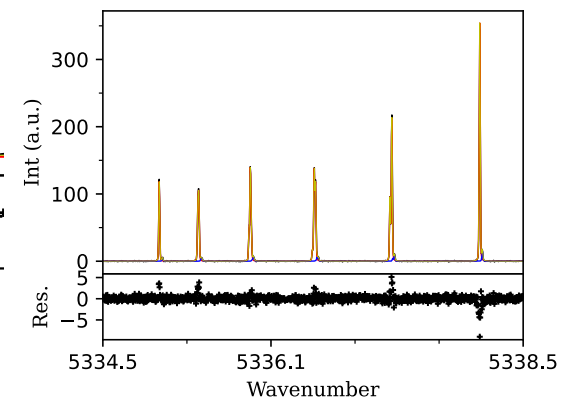
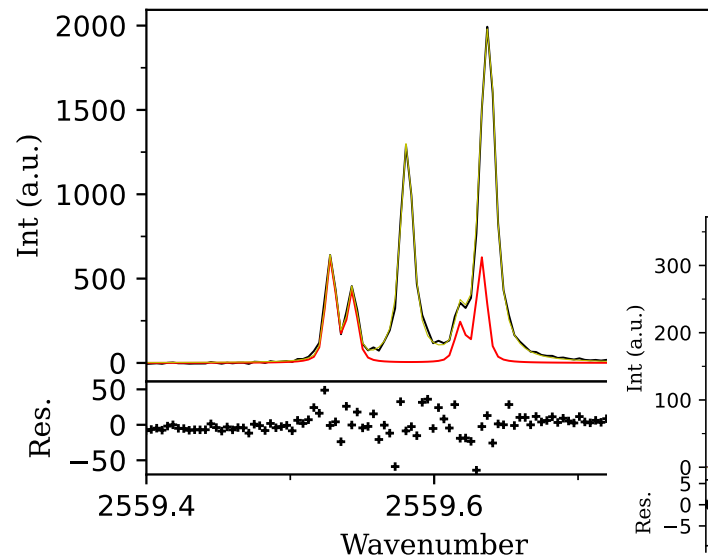
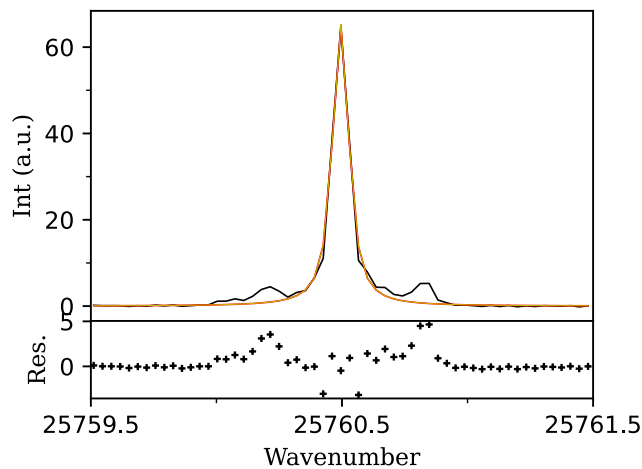
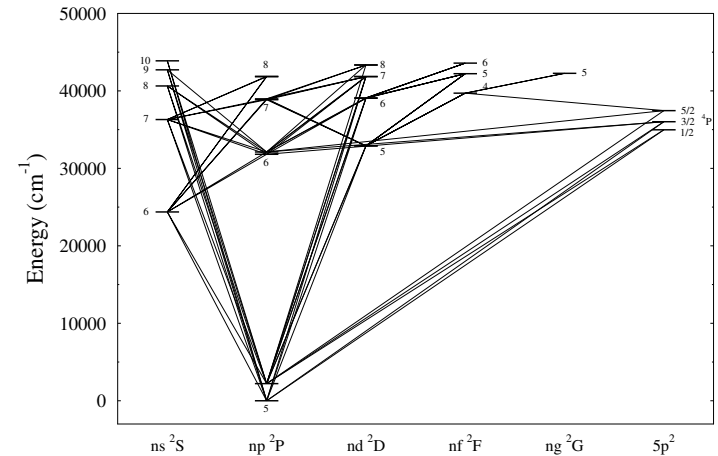
Laboratory astrophysics at Malmö University

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Laboratory astrophysics at Malmö University

- Motivation
- Laboratory measurements
- Infrared spectroscopy
- Some projects







ESO ELT

39 m main mirror

ANDES echelle
spectrograph

0.4 -1.8 μm

R = 100 000



ESO ELT

39 m main mirror

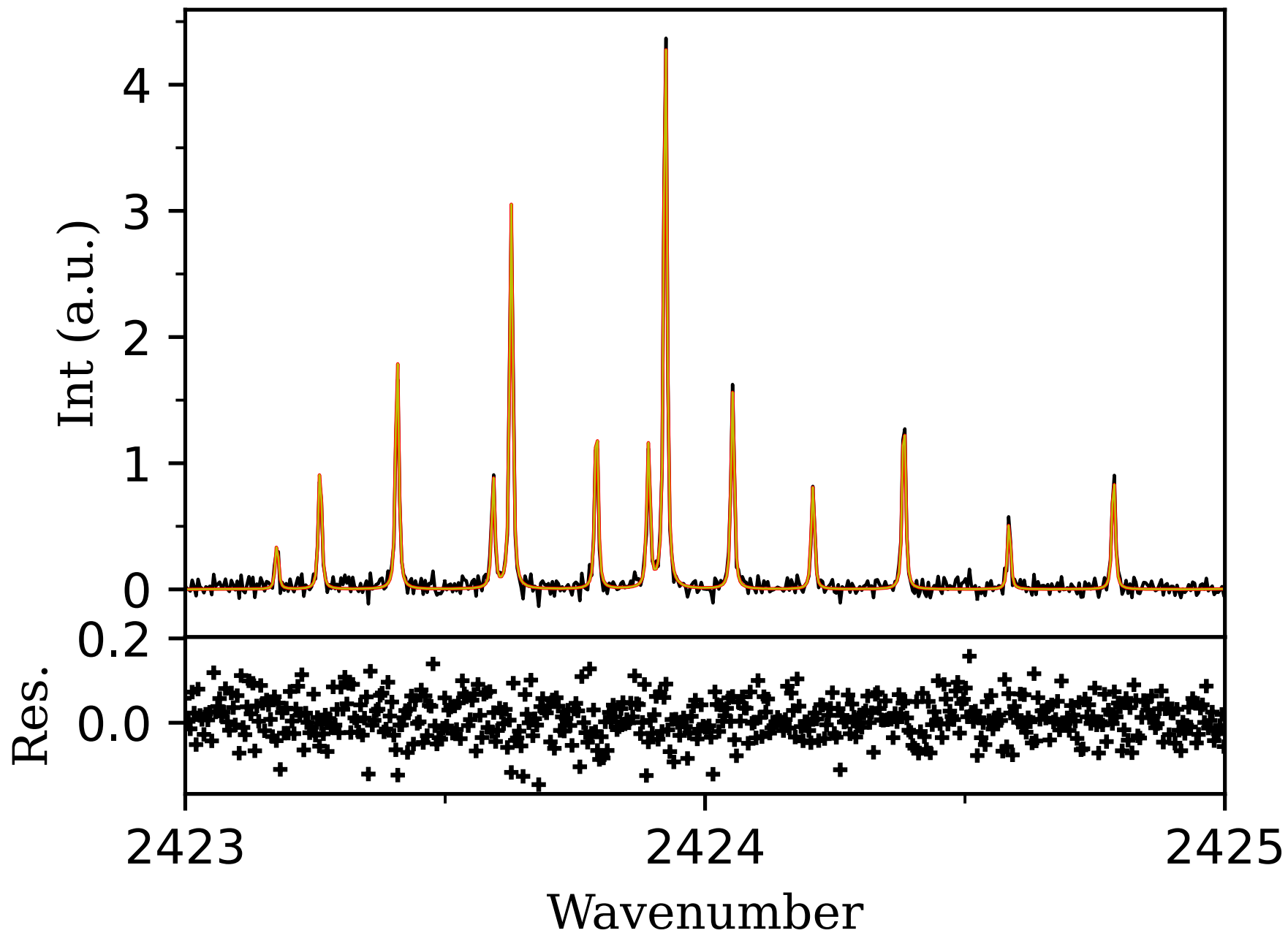
ANDES echelle
spectrograph

0.4 -1.8 μm

R = 100 000

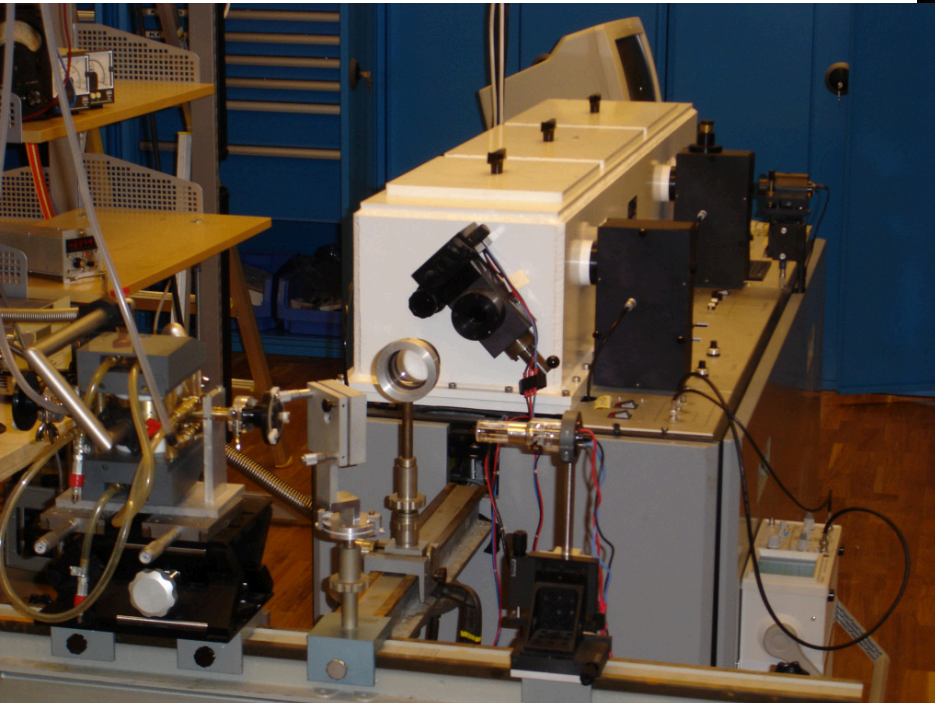
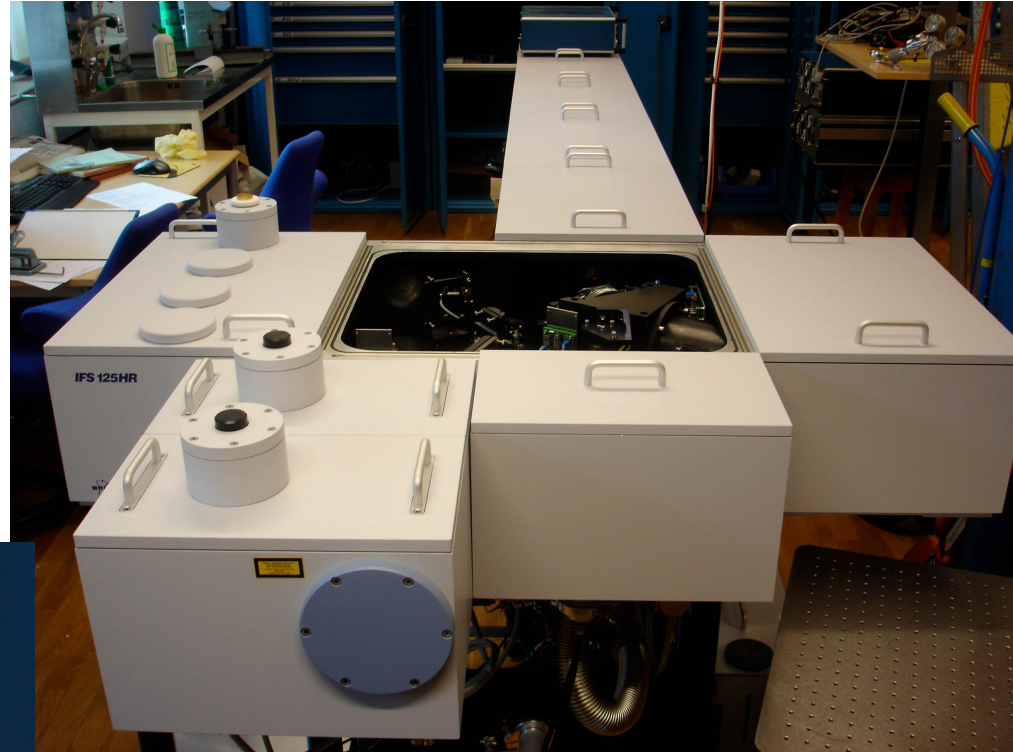
However...





Laboratory measurements

- Wavelengths
- Energy structure
- Line structure (hfs IS)
- Transition probabilities

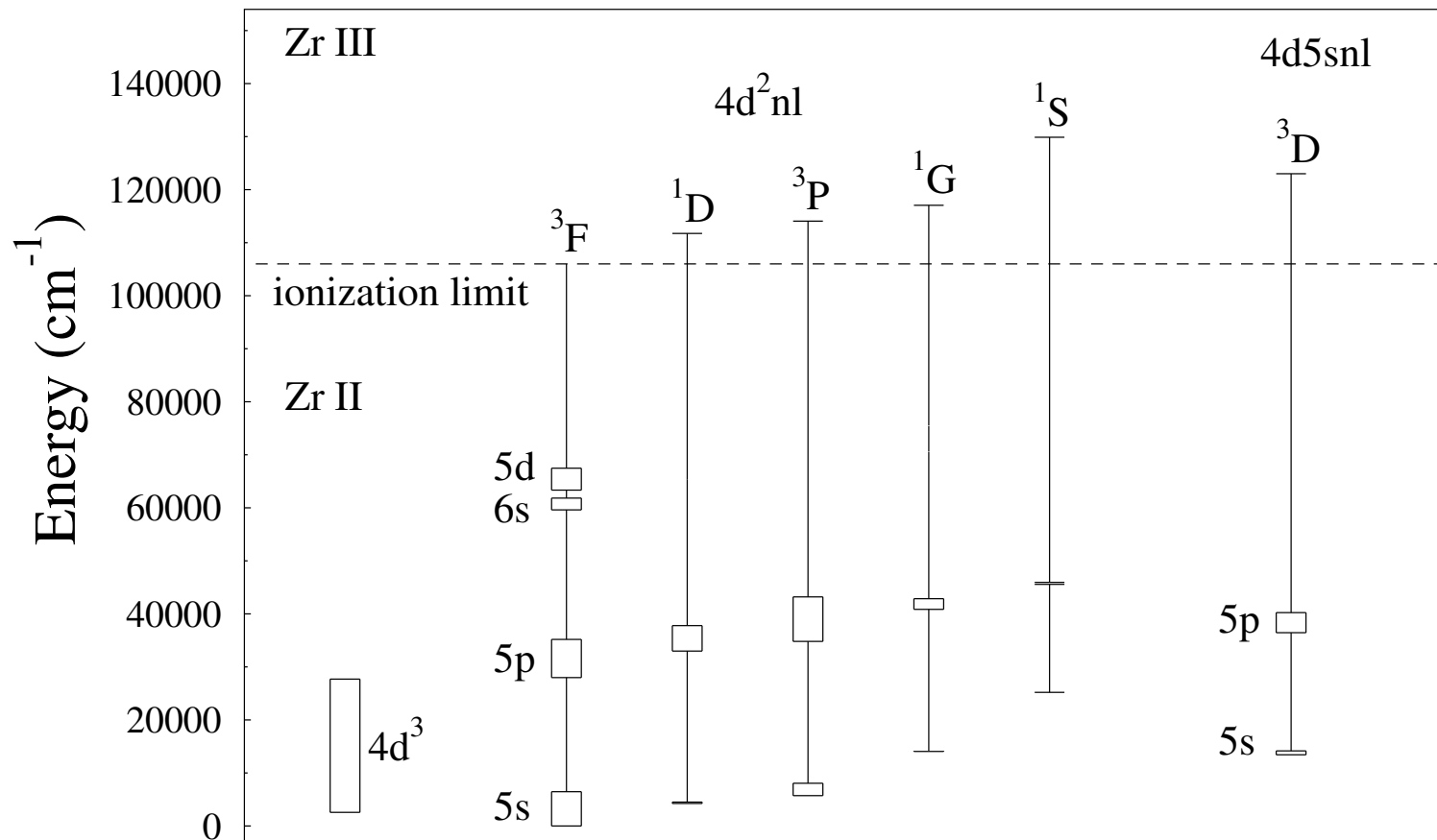


$2\ 000\text{--}50\ 000\ \text{cm}^{-1}$

$5\ \mu\text{m} - 2000\ \text{\AA}$

$R \sim 10^6$

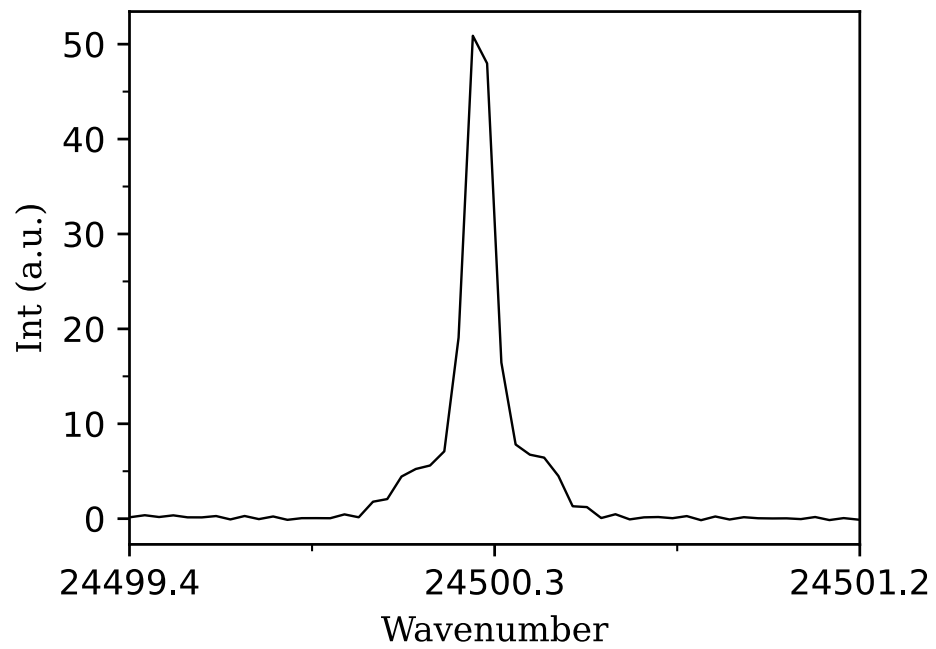
Wavelengths and energies



Partial energy level diagram of Zr II

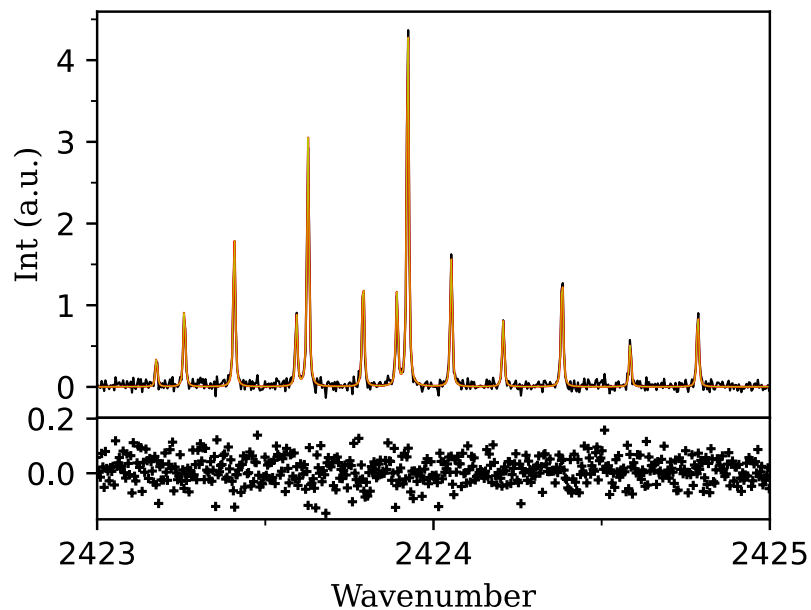
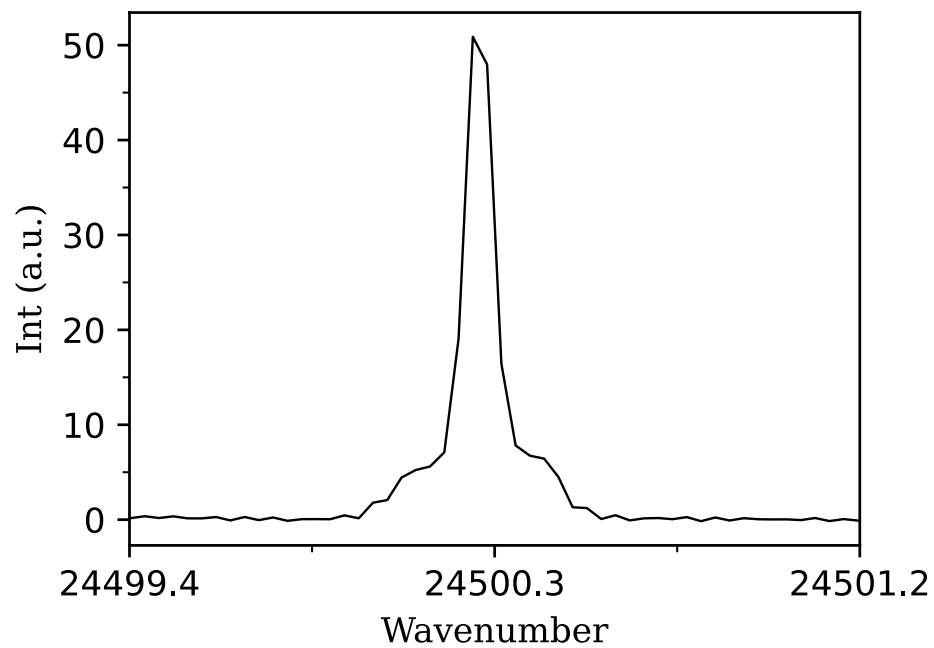
Line structure

Isotopic shift in hafnium



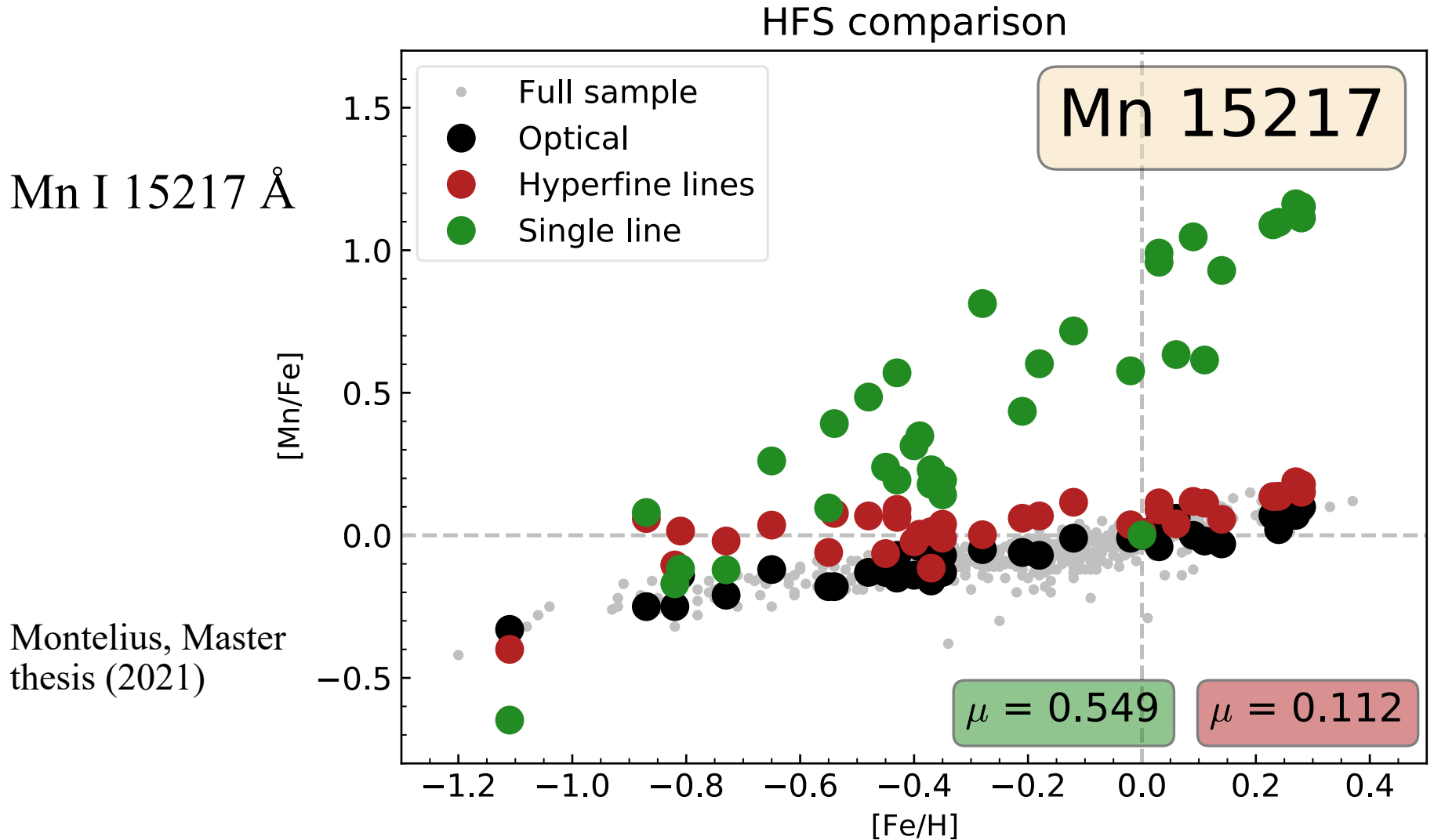
Line structure

Isotopic shift in hafnium



Hyperfine structure in indium

Hyperfine structure



Transition probabilities

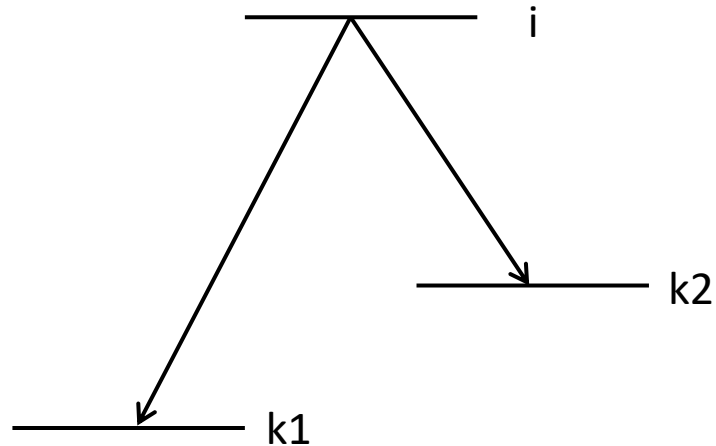
Combine relative line intensities with radiative lifetimes

branching fractions:

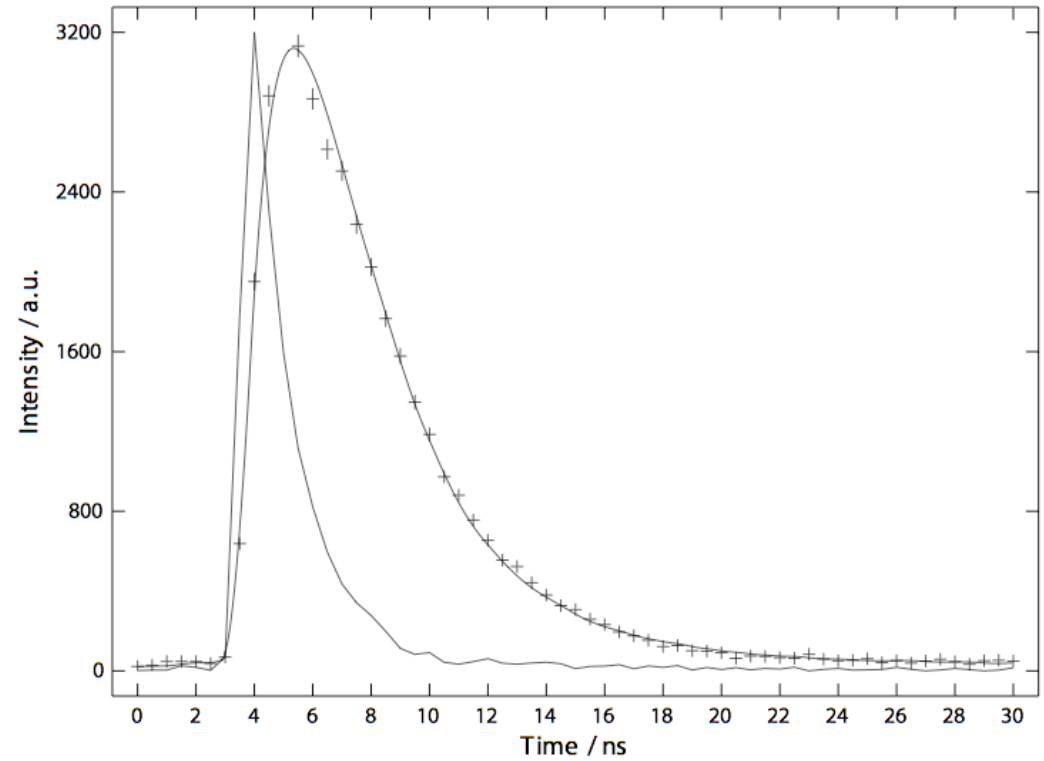
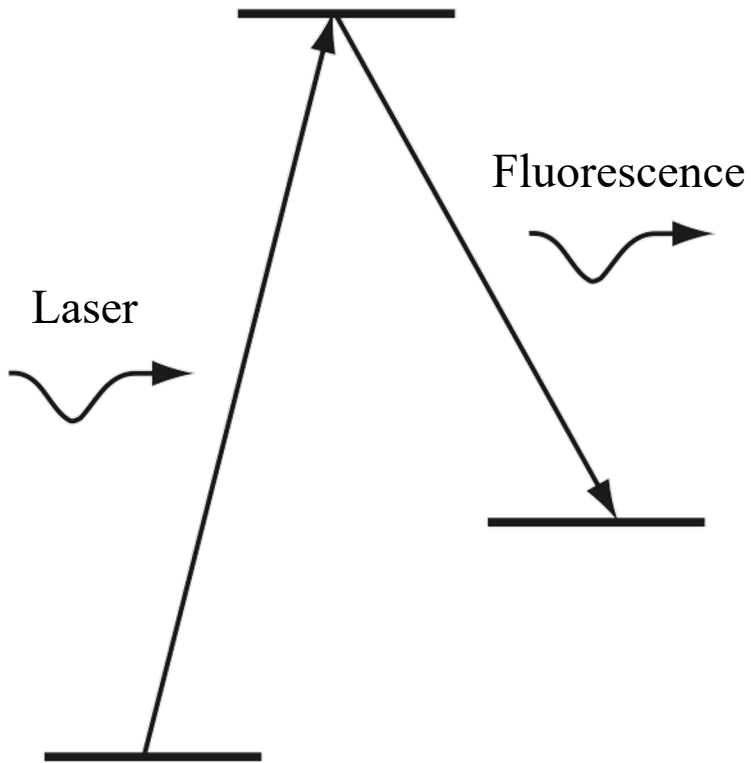
$$BF_{ik} = I_{ik} / \sum_k I$$

Transition probabilities:

$$A_{ik} = BF_{ik} / \tau_i$$



Lifetimes



Ta III (Fivet et. al. J. Phys. B: At. Mol. Opt. Phys. 41, 2008, 015702)

Why IR?

Scattering $\propto \lambda^{-4}$

Look at dusty and dense areas like the center of the Galaxy.

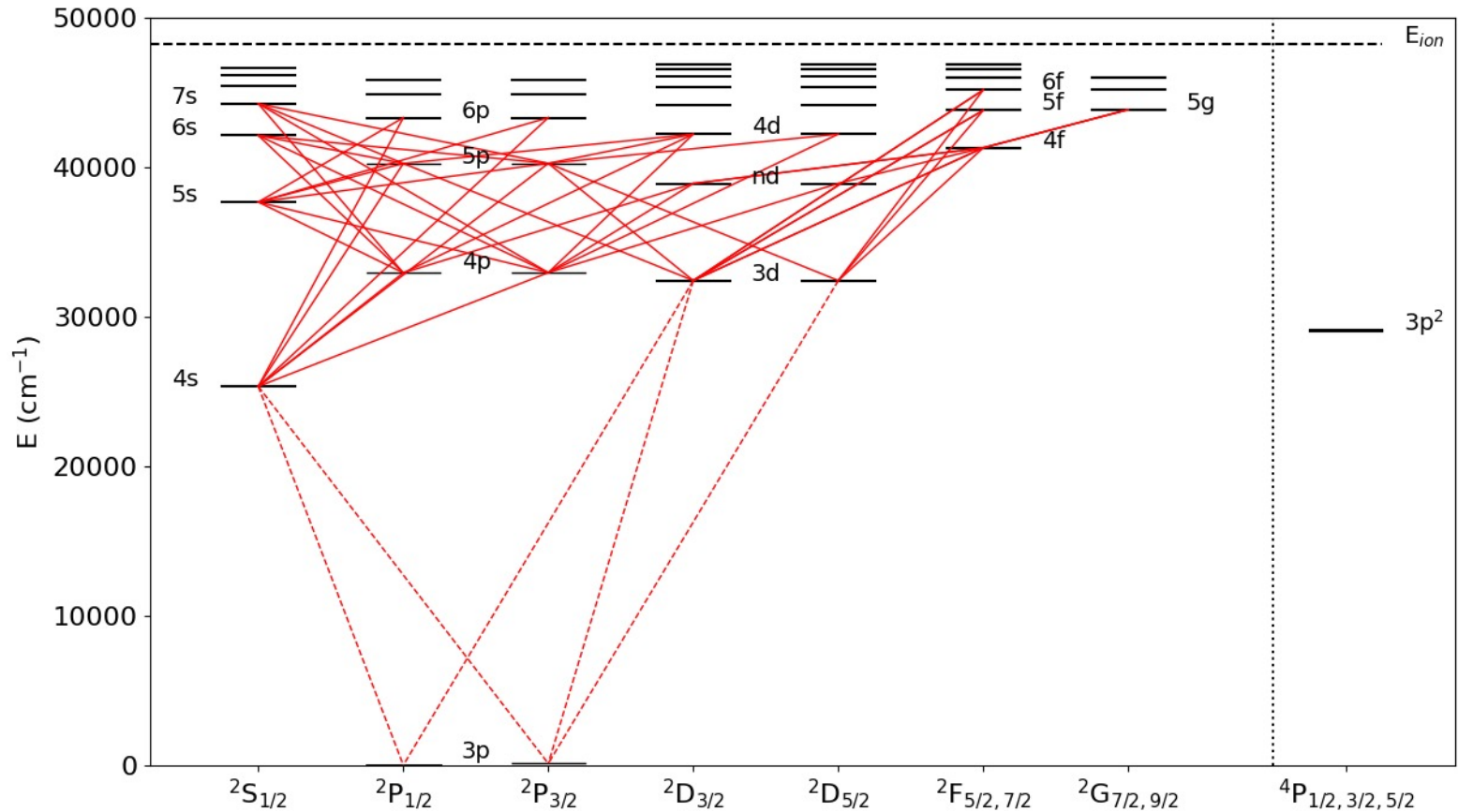
The Sun looks red at sunset and sunrise, and the sky is blue.

Cool stars to find small planets

High red shifts

Adaptive optics works better

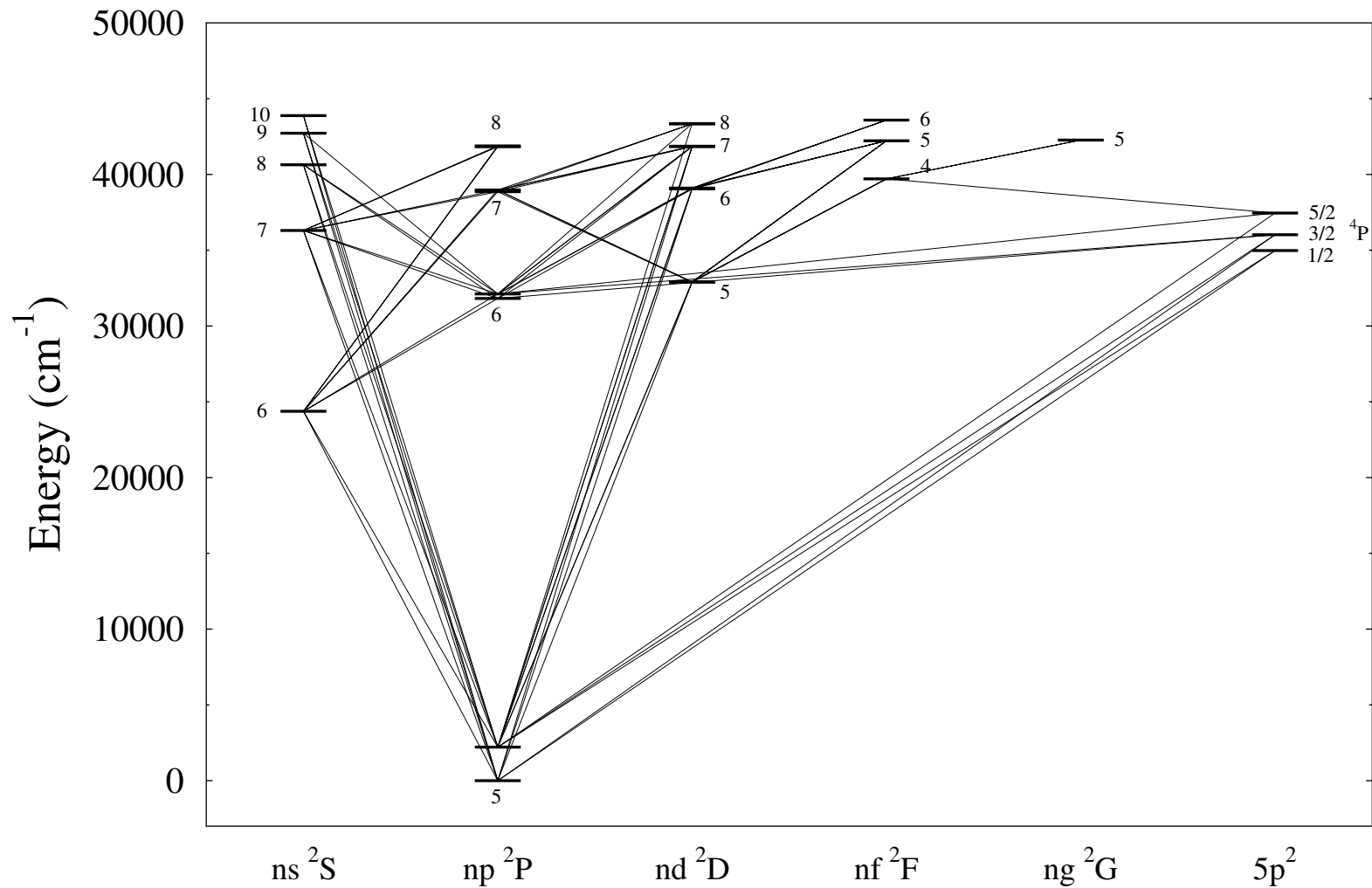
What lines fall in the IR?



Al I (Z=13)

Ground configuration $3s^2 3p$

What lines fall in the IR?

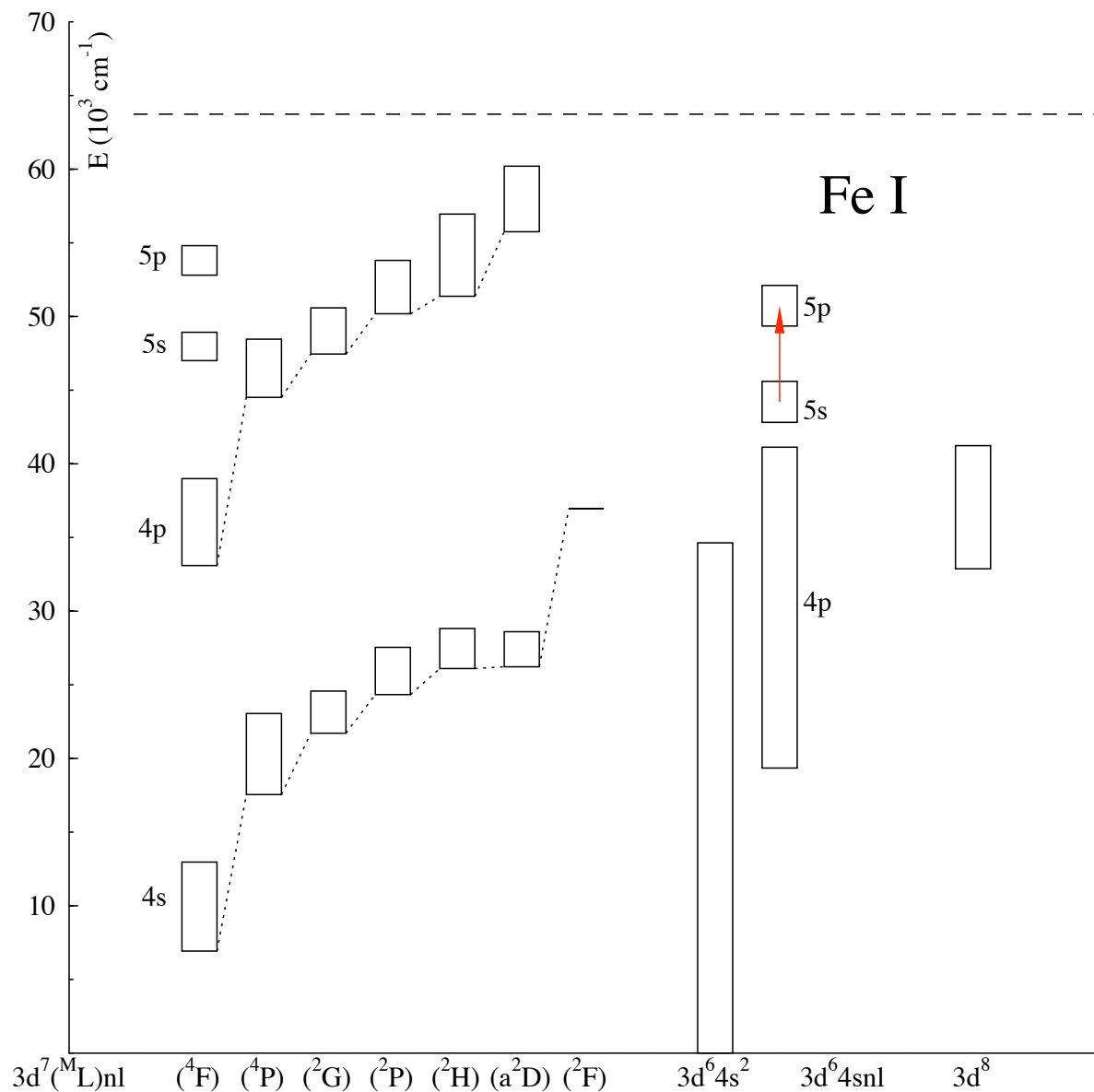


In I (Z=49)

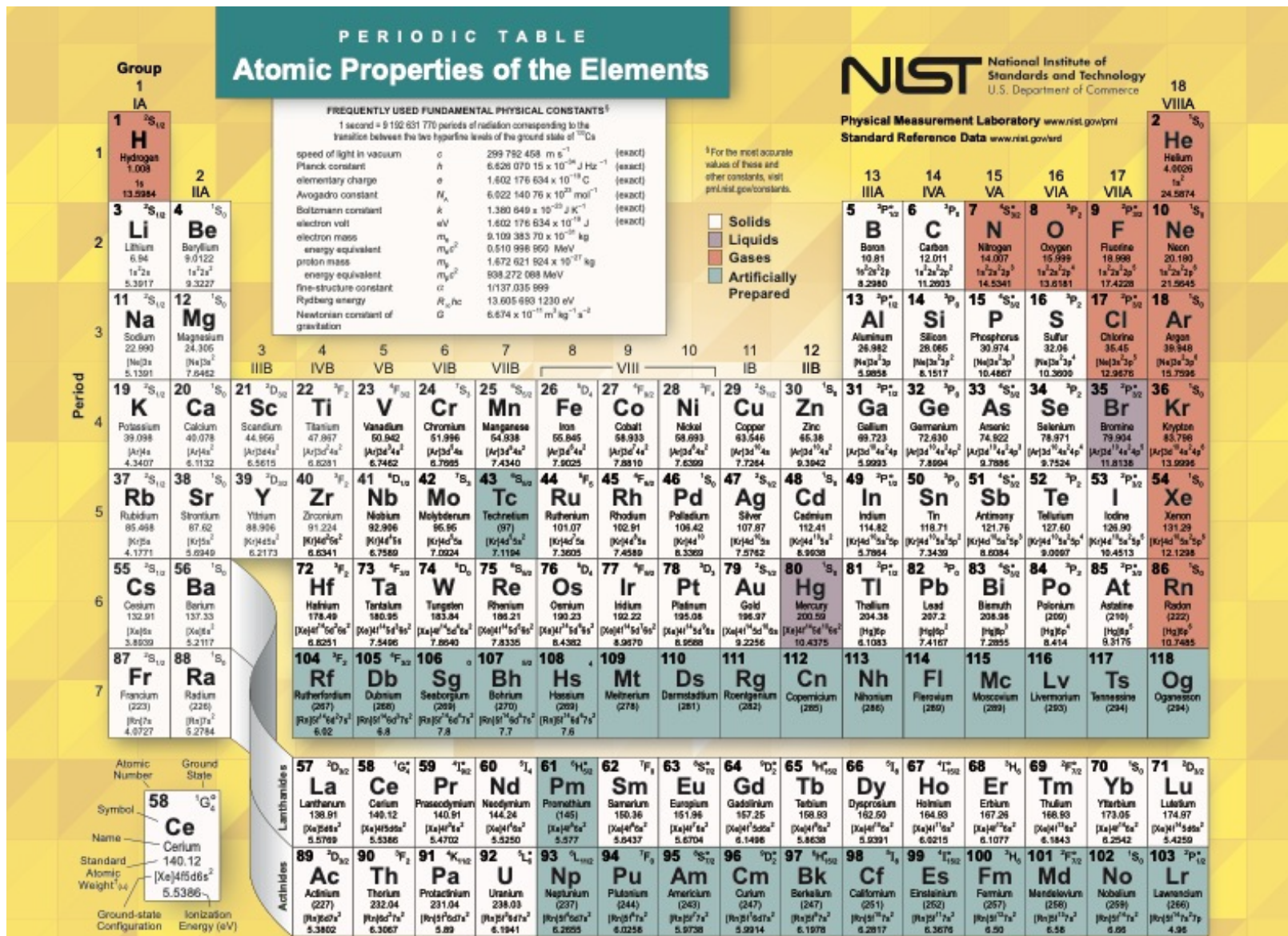
Ground configuration 5s² 5p

What lines fall in the IR?

Fe I
 Ground configuration
 $3d^6 4s^2$



What lines fall in the IR?

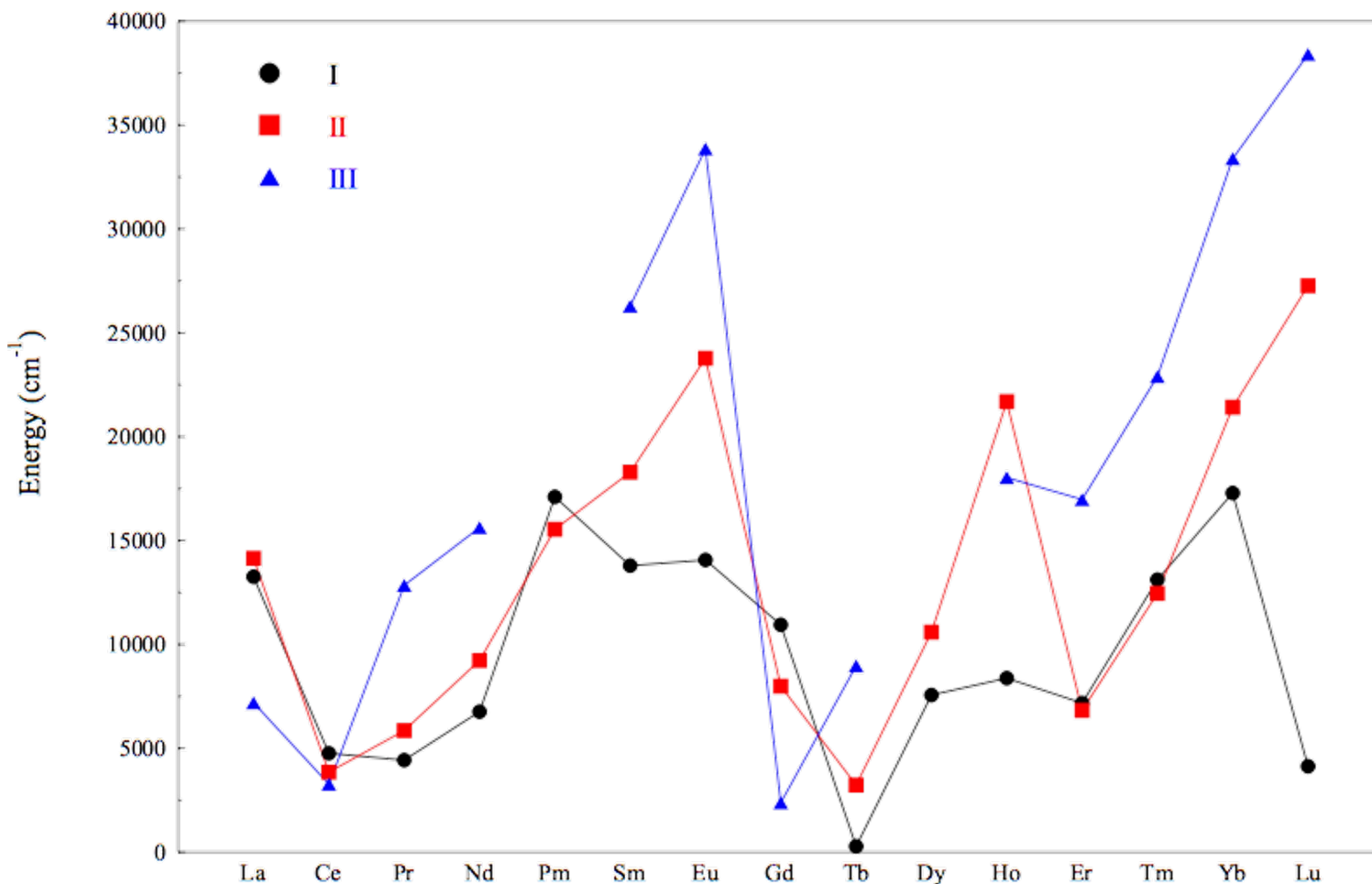


¹Based upon ¹²C. [] indicates the mass number of the longest-lived isotope.

For the most precise values and uncertainties visit ciaw.org and pml.nist.gov/data.

What lines fall in the IR?

The Lanthanides competing configurations 4f, 5d, and 6s

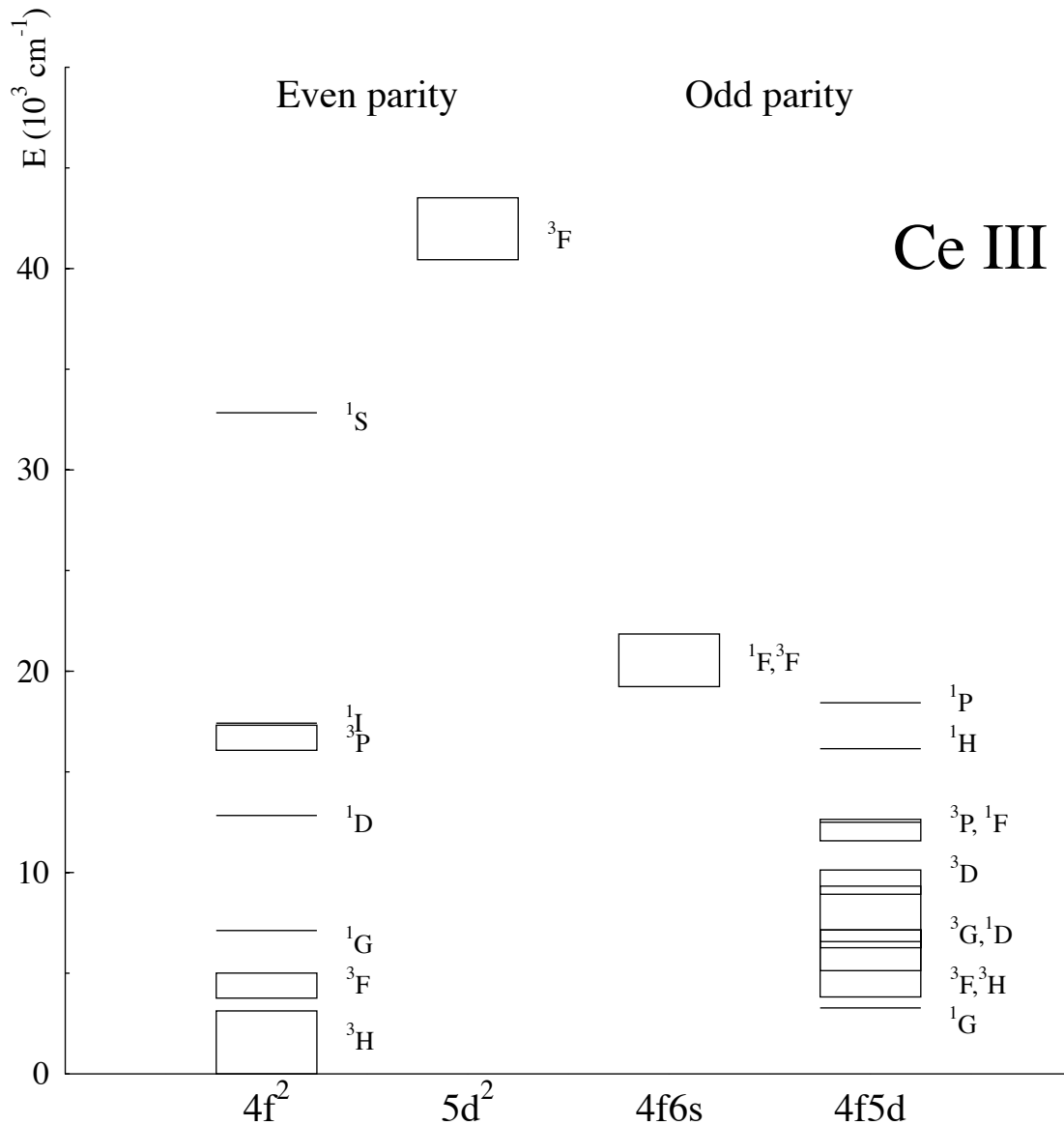


What lines fall in the IR?

The Lanthanides

Ce III

Ground configuration
4f²



Indium (Z=49)

In I

Three electron system

Ground configuration: $5s^2 5p$

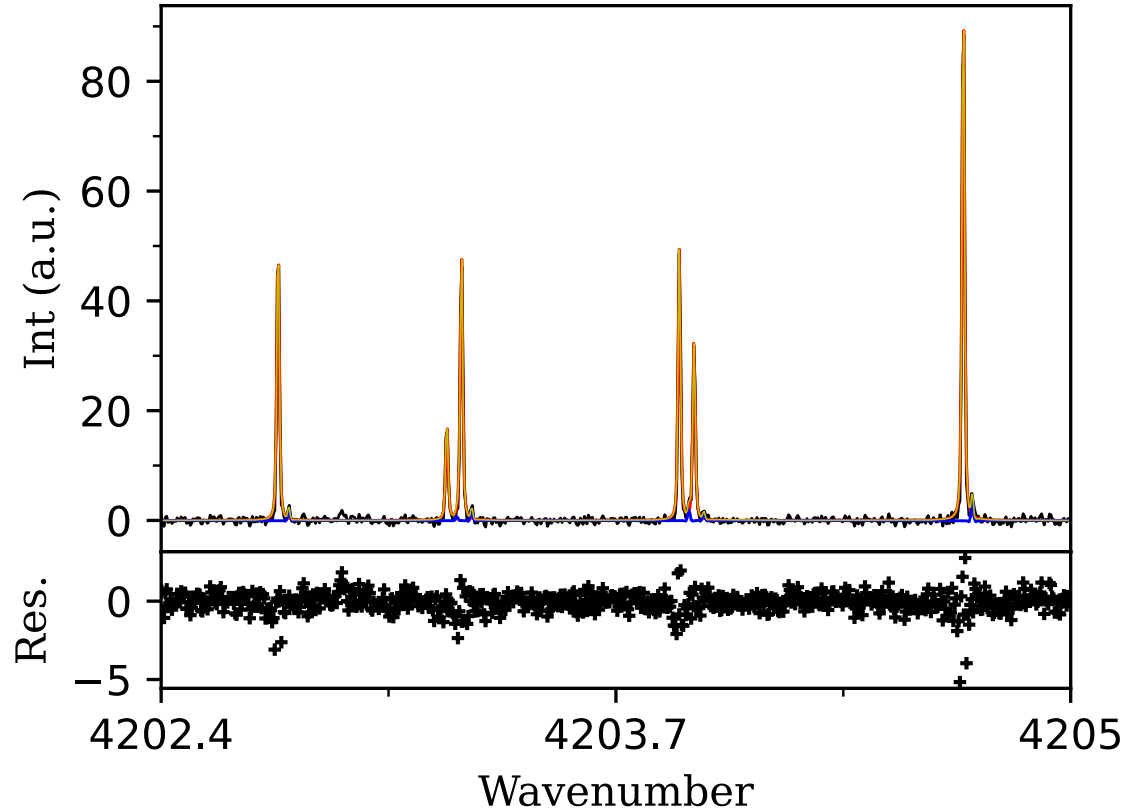
$I = 9/2$

Magnetic moment $5.54\mu_N$

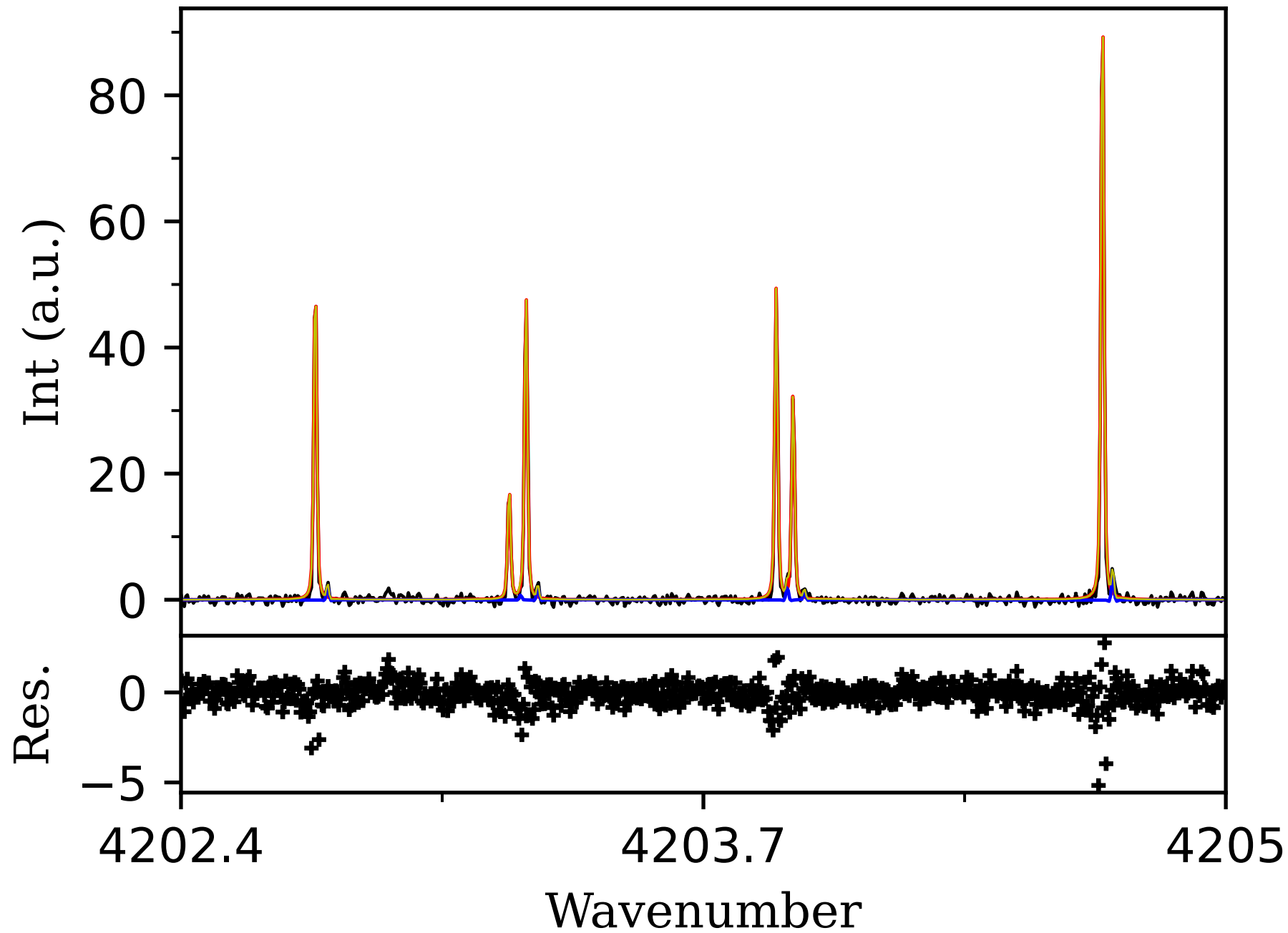
Giving rise to large hfs patterns

Two isotopes ^{113}In (4%) and ^{115}In (96%). Isotopic shift in this line is 27 mK

29 levels derived from 73 spectral lines. Hfs constants A and B for 19 levels



Intercombination line $5s^2 6p \ ^2P_{3/2} - 5s 5p^2 \ ^4P_{5/2}$

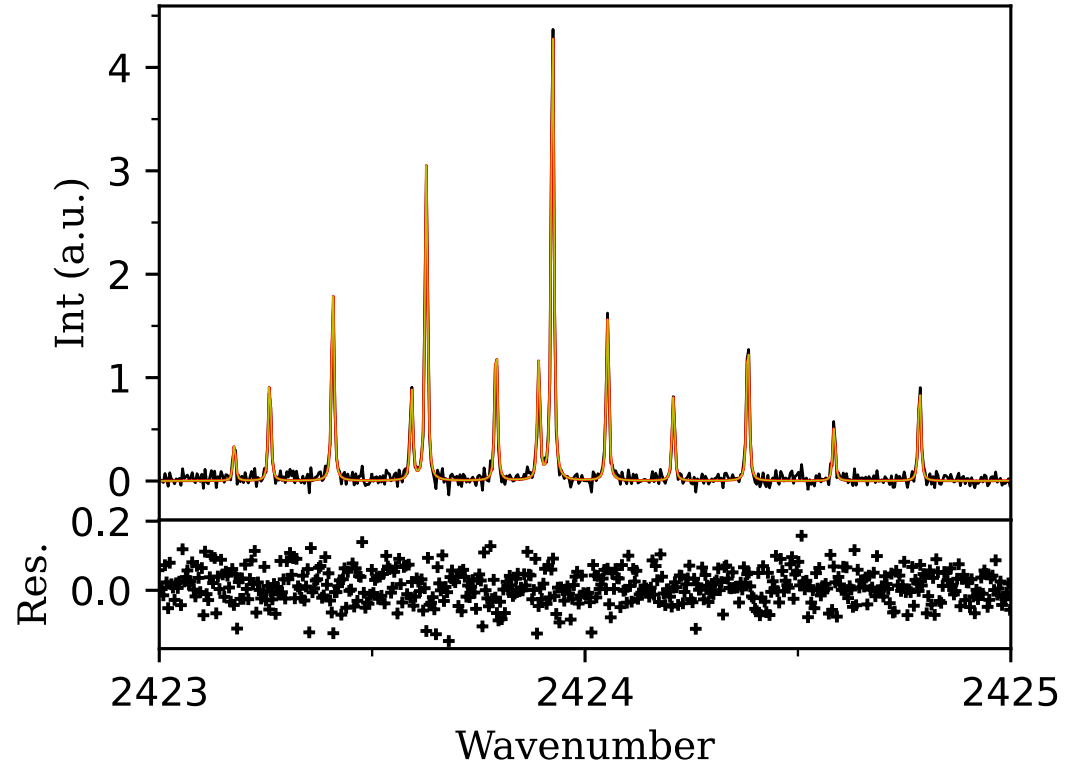


Indium ($Z=49$)

In II

Two electron system. Ground configuration $5s^2$

37 spectral lines used to determine hfs constants A and B for 21 levels.



$5s6d\ ^3D_3 - 5s7p\ ^3P_2$

Hafnium ($Z=72$)

Hf II

Three electron system.

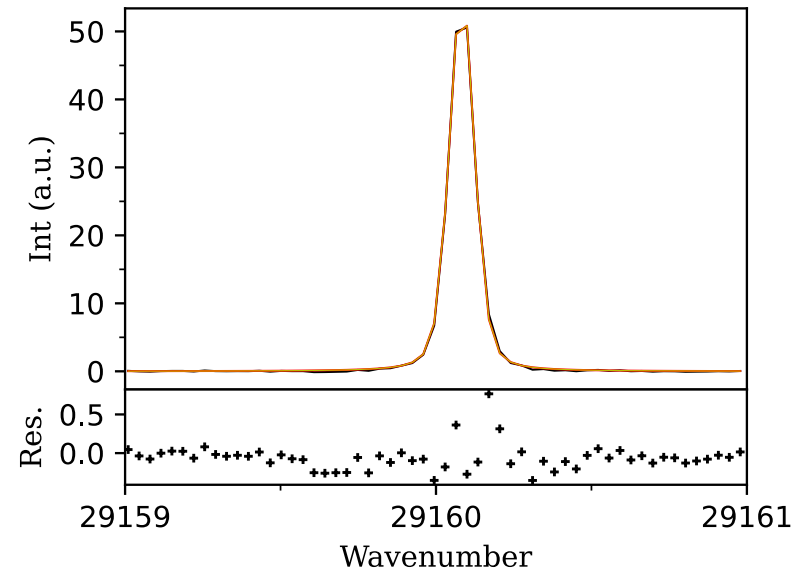
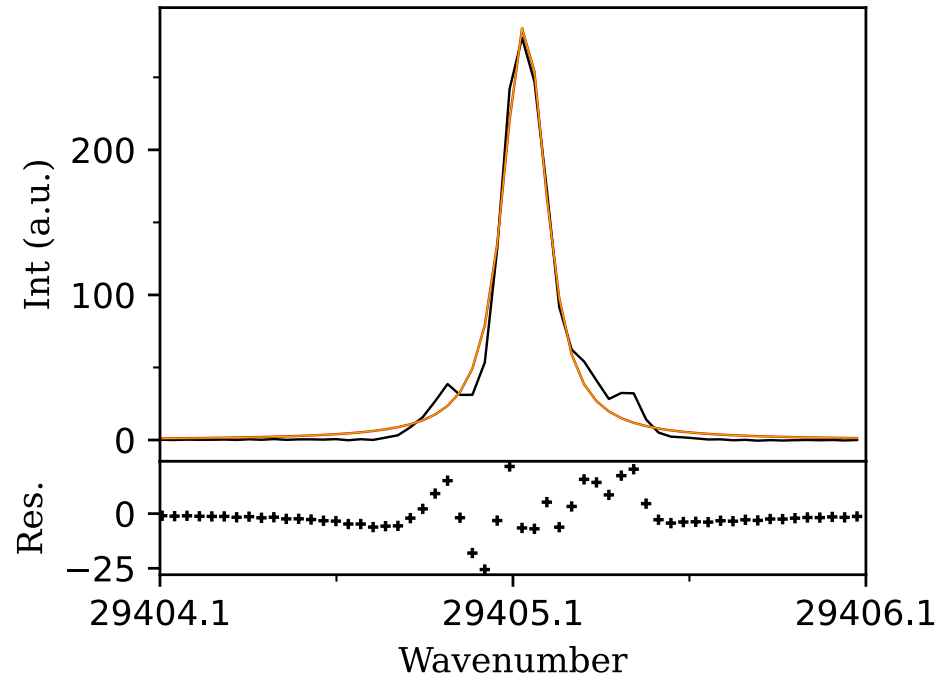
Ground configuration $5d6s^2$

Hafnium has six naturally occurring isotopes (two odd)

FTS spectra from $2000 - 50\,000\text{ cm}^{-1}$ ($5000 - 200\text{ nm}$)

10 000 spectra lines

133 levels ...



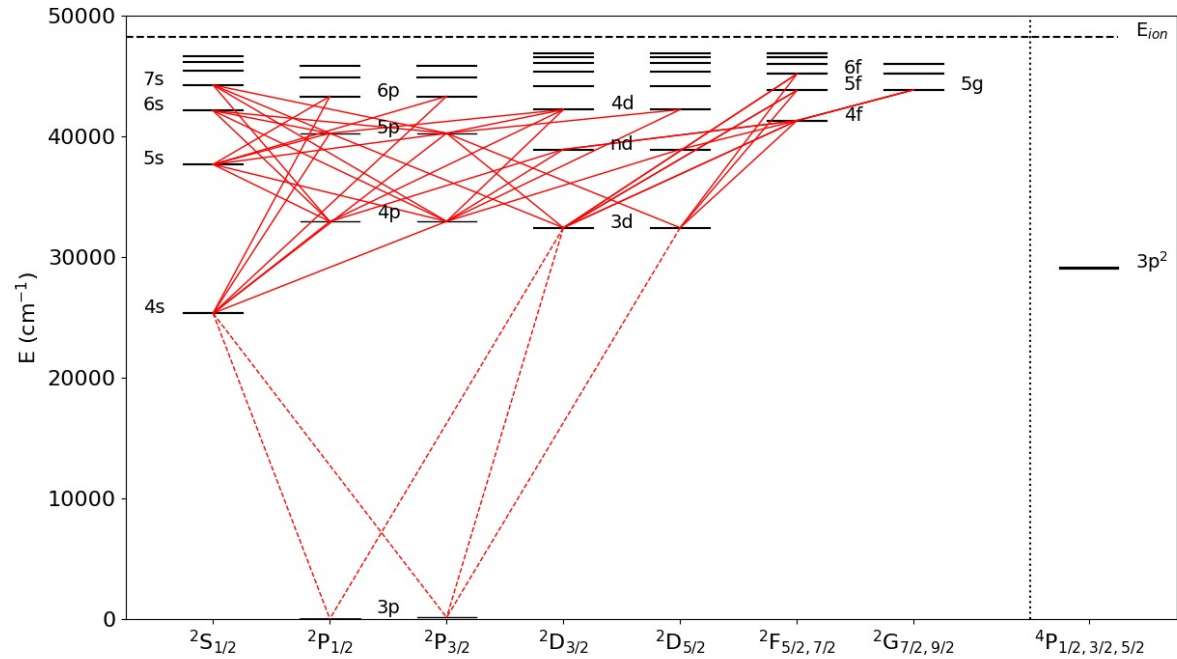
Aluminium (Z=13)

Al I

Ground configuration $3s^2 3p$

Oscillator strengths for 12 lines in the NIR and optical spectral regions, with an accuracy between 2 and 11%

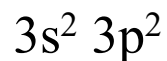
Branching fractions for an additional 16 lines



Silicon Z=14

Si I and Si II

Ground configuration

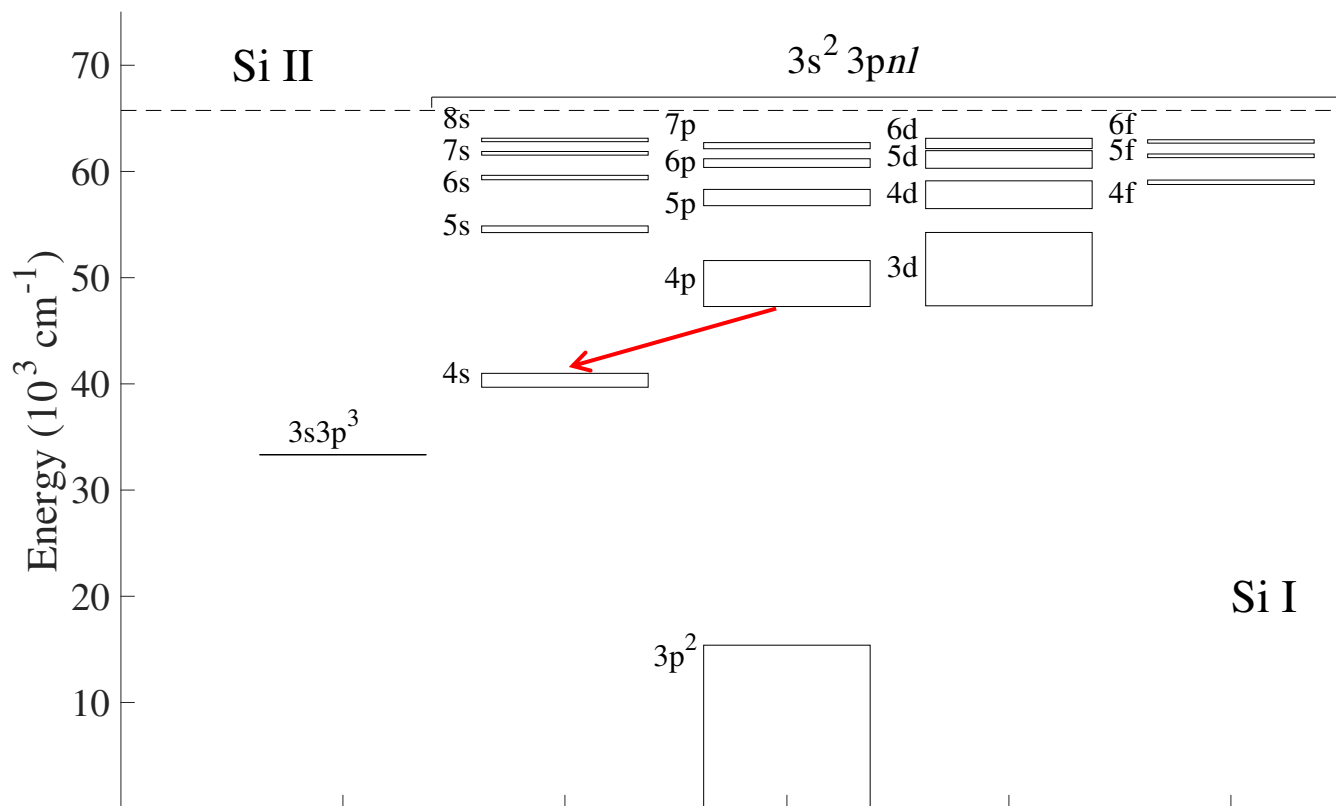


We report 17 log *gfs* for $3s^2 3p4s$ - $3s^2 3p4p$ IR-lines.

Derived by combining experimental BFs and lifetimes calculated with GRASP2K.

1500 Si I and 500 Si II theoretical log *gfs*

Accepted by A&A



Zirconium (Z=40)

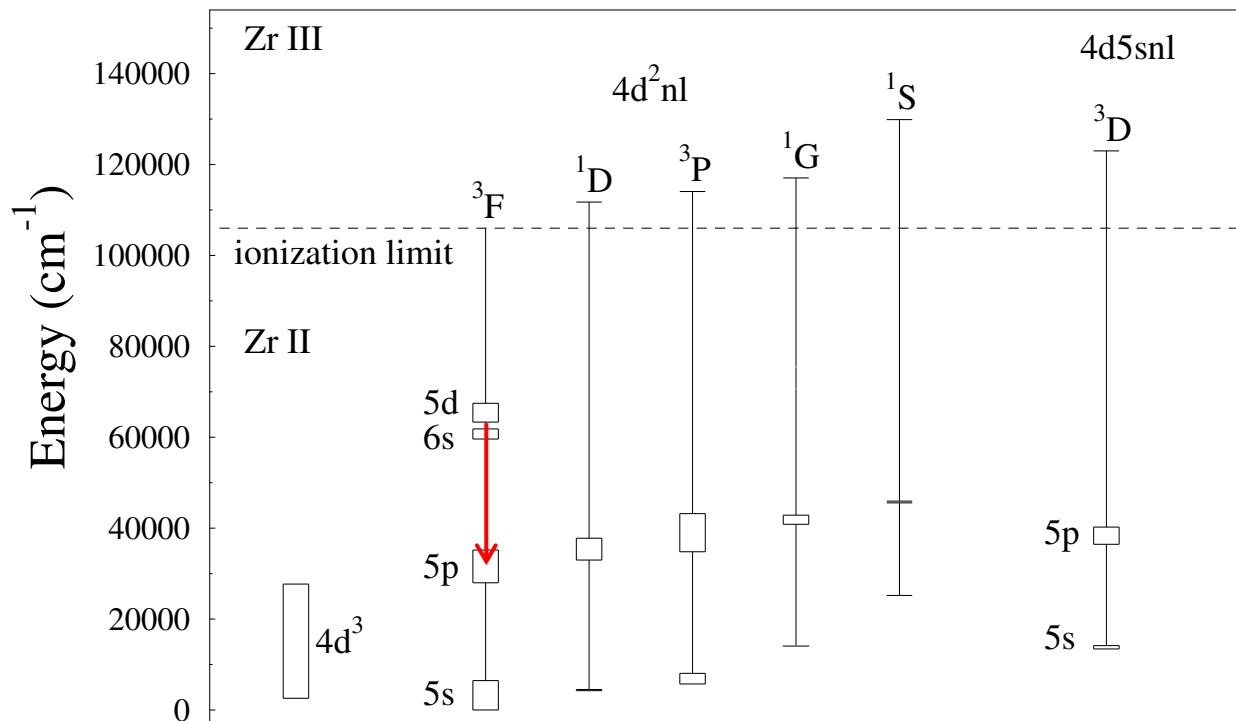
Zr II (Poster Burheim et. al.)

Three electron system
(Zr II)

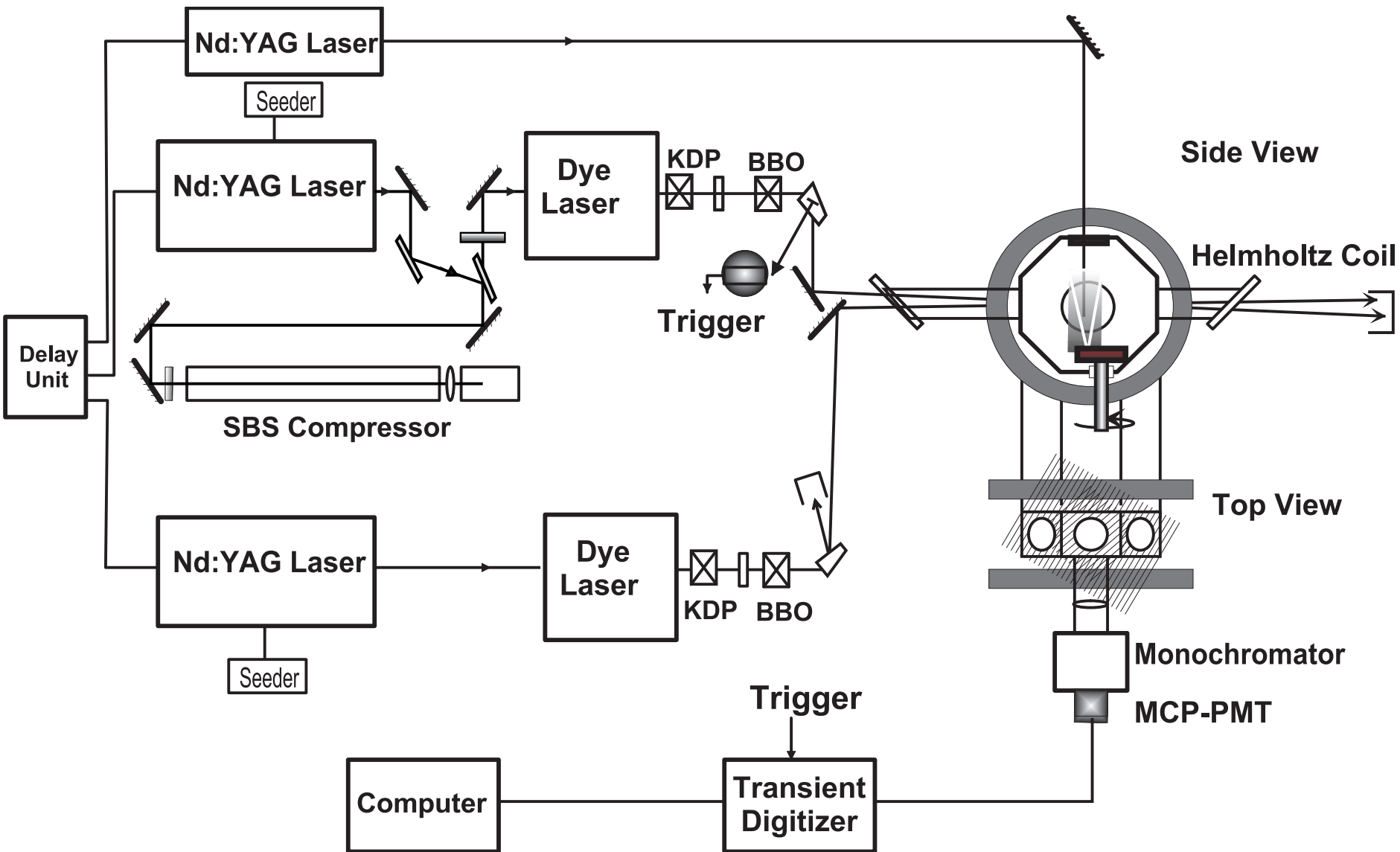
Ground configuration
 $4d^3$

New energy levels 6s
and 5d

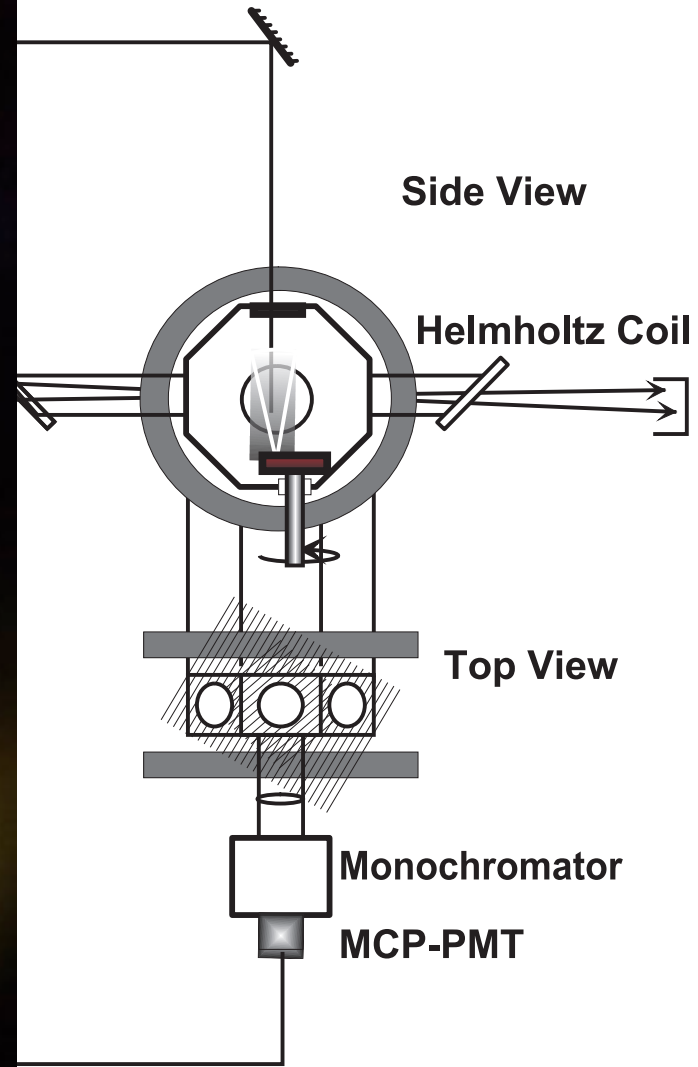
Wavelengths and BFs
for 106 lines from 20
levels



Lifetimes



Lifetimes



Laboratory astrophysics at Malmö University

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